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PFAS Removal from

Hard-to-treat Waste Streams

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Center for Environmental Solutions & Emergency Response (CESER)





PFAS



Monday, August 14, 2023 from 8:40 AM - 9:00 AM Room 3001, West Bldg. - Moscone Center

Room 3008, West Bldg. - Moscone Center





pubs.acs.org/journal/estlcu

Population-Wide Exposure to Per- and Polyfluoroalkyl Substances from Drinking Water in the United States

David Q. Andrews* and Olga V. Naidenko

Estimated population-wide exposure to PFOA and PFOS from drinking water in the United States

Letter







pubs.acs.org/estwater

The Price of Really Clean Water: Combining Nanofiltration with Granular Activated Carbon and Anion Exchange Resins for the Removal of Per- And Polyfluoralkyl Substances (PFASs) in Drinking Water Production

Vera Franke,* Malin Ullberg,* Philip McCleaf, Maria Wålinder, Stephan J. Köhler, and Lutz Ahrens



| Σ_{11} PFASs [ng L ⁻¹] - | 580 | | | | 1.6 |
|--|------|----|-----|---|-------------|
| Hardness [°dH]- | 69 | | | | 6.5 |
| Uranium [µg L ⁻¹]- | 170 | | | | 1 |
| Br ⁻ [µg L ⁻¹]- | 460 | | | | 120 |
| DOC [mg L ⁻¹]- | 16 | | | | 0.5 |
| Cl ⁻ [mg L ⁻¹]- | 110 | | | | 26 |
| COD [mg L^{-1}]- | 6.9 | | | | 0.5 |
| Na ⁺ [mg L ⁻¹]- | 73 | | | | 21 |
| Ca ²⁺ [mg L ⁻¹]- | 380 | | | | 42 |
| Mg ²⁺ [mg L ⁻¹]- | 69 | | | | 3 |
| F ⁻ [mg L ⁻¹]- | 4.7 | | | | 0.44 |
| SO ₄ ²⁺ [mg L ⁻¹]- | 210 | | | | 1 .5 |
| HCO ₃ ⁻ [mg L ⁻¹]- | 1200 | | | | 150 |
| Mn²+ [µg L⁻¹]- | 41 | | | | 0.78 |
| NO ₃ ⁻ [mg L ⁻¹]- | 9 | | | | 3.1 |
| | 1(| 0% | 50% | 0 | 50% |

Concentrate — Permeate —



Article





Major Issues with PFAS Removal/Degradation:

- Over 10000 chemicals in the market.
- Highly water soluble
- Low concentration, when compared to other background constituents in water.
- C-F bond is shortest and strongest bond in nature.











Detect • Adsorb • Destroy









www.acsmaterialsletters.org

Best Practices for Evaluating New Materials as Adsorbents for Water Treatment

Mohamed Ateia,* Damian E. Helbling, and William R. Dichtel*







Adsorption mechanisms of PFAS



Factors that affect PFAS removal:

- Hydrophobic interactions.
- Electrostatic interactions.
- Pore size & structure.







Cite This: Environ. Sci. Technol. Lett. 2019, 6, 688–695



Efficient PFAS Removal by Amine-Functionalized Sorbents: Critical Review of the Current Literature

Mohamed Ateia,**^{†,‡,||} Alaaeddin Alsbaiee,^{§,||} Tanju Karanfil,[†] and William Dichtel[‡]







Q: Why are amine-containing sorbents efficient for PFAS removal?

A: The interplay between electrostatic interactions,

hydrophobic interactions, and pore size.



Ateia et al., 2019; Ateia et al., 2018



New Optimized Cyclodextrin Adsorbents



✓ Modular Synthesis







New Optimized Ionic Fluorogels







Prof. Frank Leibfarth

Irene Mulloy Manning

Manning et al., Angewandte Chemie 134.41 (2022)



New Optimized granular adsorbent





Not EPA data – Results were provided by the company form a third-party evaluation.











TWO Postdoc Positions

Thank you! Contact: ibrahim.mohamed@EPA.GOV

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